High Diastereofacial Selectivity in the 1,3-Dipolar Cycloaddition of Chiral Azomethine Ylides

Pierre Deprez, Jacques Royer and Henri-Philippe Husson

Institut de Chimie des Substances Naturelles C.N.R.S., 91198 Gif-sur-Yvette Cedex, France

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Abstract : Excellent diastereofacial and endo/exo stereoselectivities have been obtained in cycloaddition of the chiral azomethine ylide generated from 4-phenyloxazolidine acetic acid (-)-8-phenylmenthyl ester 1e.

The 3+2 cycloaddition reactions of azomethine ylides are very well documented and widely studied. Surprisingly, very little work has been concerned with facial diastereoselectivity and furthermore successes in this area remain modest or are limited to isolated instances¹ (intramolecular processes for example).

We recently introduced a new facile method for the generation of ylides under very mild conditions using N-methoxycarbonylmethyl-4-phenyloxazolidine $1a^2$. Cycloaddition to various activated dipolarophiles occurred in high yield within 4 h at - 78°C. This represented one of the first examples of a chiral azomethine ylide and we undertook a study of the diastereoselectivity of the cycloaddition³. Using N-phenylmaleimide as dipolarophile, it was shown that cycloaddition proceeded with complete *exo* selectivity on the stabilized U-shaped ylide Y_1 (Scheme 1) derived from 1a but without any facial selectivity (two cycloadducts 2a and 3a are formed in a 48:52 ratio).



This low diastereofacial selectivity was not improved on by modification of substituents R^1 and R^2 on the oxazolidine ring of $1.^3$ A possible explanation is that in the *exo* mode of cycloaddition the chiral center cannot interact with the dipolarophile because of its remote position. We thus proposed the introduction of a chiral group in the ester moiety and we present herein our preliminary results obtained in cycloaddition reactions of such derivatives with N-phenylmaleimide.

Esters 1b-e were easily synthesized in good yields from commercially available (+)-menthol, (-)-menthol and (-)-8-phenylmenthol (Scheme 2). R-(-)-Phenylglycinol and 2-amino-2-methyl propanol were N-monoalkylated with menthyl bromoacetate (obtained by treatment of the appropriate menthol with bromoacetic acid in the presence of pyridine) in THF with 1 eq of iPr₂NEt followed by treatment with bromomethyl methyl ether in the same medium or by condensation with paraformaldehyde in refluxing toluene (Table 1).





Table 1 : Preparation of chiral menthyl esters 1.

Menthol*	R ¹	R ²	<u>Oxazolidine</u>	Overall yield (%)	[α] _D ²⁰	
(+)-menthol	CH3	CH3	16	45	+57 (c 1, CHCl ₃)	
(+)-menthol	Ph	Н	1 c	58	-41 (c 5, CHCl3)	
(-)-menthol	Ph	Н	1 d	50	-144 (c 1.2, CHCl3)	
(-)-8-phenylmenthol	Ph H le		77	-59 (c 3, CHCl ₃)		



Table 2: Isomeric ratio of different cycloadducts of N-phenylmaleimide with different chiral esters 1.

	R	R1	R ²	Overall yield	H-2/H-3 cis cxo		H-2/H-3 trans endo		exo/endo	d.e.
					2	3	4	5		
(a)	Me	Ph	Н	85	48	52	-	-	100:0	4
(b)	(+)-Menthyl	Me	Mc	52	55	-	45	-	55:45	-
(c)	(+)-Menthyl	Ph	Н	83	84	9	7	-	93:7	80
(d)	(-)-Menthyl	Ph	н	79	20	80	-	-	100:0	60
(e)	(-)-8-Ph Menthyl	Ph	Н	86	-	≥98	-	-	100:0	≥95

We first used ester 1b wich had no chiral group on the oxazolidine ring and it can be seen from Table 2 that two cycloadducts 2b and 4b⁴ (Scheme 3) were obtained in a 45:55 ratio.⁵ We deduced that major compound 2b arose from an *exo* cycloaddition whereas 4b was an *endo* cycloadduct. This result was in agreement with previous results from this laboratory ³ showing that small R¹ and/or R² groups led to a fair *endolexo* selectivity. We then used oxazolidines derived from R-(-)-phenylglycinol in the hope of obtaining only *exo* cycloadducts. The two diastereomeric esters 1c and 1d obtained respectively from (+)- and (-)-menthol were investigated. The cycloaddition of *N*-phenylmaleimide with 1d (derived from (-)-menthol) gave an 80% yield of two *exo* adducts ⁴ (2d and 3d) with 60% d.e.⁵ showing the determinant role of the menthyl group. On the other hand, ester 1c (derived from (+)-menthol) gave a very good facial diastereoselectivity : d.e. = 80% [In this instance the cycloaddition was not completely *exo* : the *exo-endo* ratio was 93:7].

Although this process is not strictly speaking a double induction - the two chiral groups are placed on the same reactive species - it clearly proceeds on the same principle.

We finally tried oxazolidine 1e derived from (-)-8-phenylmenthol which has proved to be a much more effective chiral directing group than menthol, in the Diels-Alder reaction of menthyl acrylates for instance.⁶ We were pleased to find that a single cycloadduct $3e^{7,8}$ was obtained in 86% yield from the reaction of 1e with *N*-phenylmaleimide. This indicates that, in this case, one face of the ylide is completely masked by the phenyl ring of the 8-phenylmenthyl group. The stereochemistry of 3e implicates an *exo* transition state. To explain the stereochemical course of the reaction, we propose the model shown on Figure. Determination of the absolute configuration of cycloadduct 3 will give us new elements to confirm this model. Work is in progress in this field and will be reported later.



We have thus designed and synthesized a new stable synthon $1e^9$ which is easily transformed to an azomethine ylide constrained in a U-shaped conformation, and bearing two chiral moieties. The first chiral center borne by the nitrogen atom forces the exclusive *exo* addition of dipolarophile while the second chiral center on the ester group permits diastereofacial selectivity.

Oxazolidine le was treated with a variety of dipolarophiles to check the generality of the reaction. Cycloadditions with dimethyl acetylenedicarboxylate and 2,2-dimethyl cyclopentene-1,3-dione occurred in good yield (respectively 81% and 56%) giving in each case a single product, but dimethylmaleate did not react.

REFERENCES AND NOTES

 Examples using chiral ylides : a) Padwa, A.; Chen, Y. Y.; Chiacchio, U.; Dent, W. Tetrahedron 1985, 41, 3529. b) Negron, G.; Chastanet, J.; Roussi, G. Personal communication (J. Org. Chem. submitted).
c) Garner, P.; Sunitha, K.; Shanthilal, T. Tetrahedron Lett. 1988, 29, 3525. d) Rouden, J. Ph.D Thesis (Paris XI, 1990). e) Anslow, A.S.; Harwood, L.M.; Phillips, H.; Watkin, D. Tetrahedron Asymm. 1991, 2, 169.
Examples using chiral dipolarophiles : f) Takahashi, T.; Kitano, K.; Hagi, T.; Nihonmatsu, H.; Koizumi, T. Chem. Lett. 1989, 597. g) Kanemasa, S.; Yamamoto, H. Tetrahedron Lett. 1990, 31, 3633. h) Garner, P.; Ho, W. B. J. Org. Chem. 1990, 55, 3973. i) Barr, D. A.; Dorrity, M. J.; Grigg, R.; Malone, J. F.; Montgomery, J.; Rajviroongit, S.; Stevenson, P. Tetrahedron Lett. 1990, 31, 6569. j) Coulter, T.; Grigg, R.; Malone, J.F.; Sridharan, V. Tetrahedron Lett. 1991, 32, 5417.

Intramolecular approach : k) Takano, S.; Sugihara, T.; Satoh, S.; Ogasawara, K. J. Am. Chem. Soc. 1988, 110, 6467. l) Garner, P.; Sunitha, K.; Ho, W. B.; Youngs, W. J.; Kennedy, V. O.; Djebli, A. J. Org. Chem. 1989, 54, 2041.

- a) Rouden, J.; Royer, J.; Husson, H.-P. Tetrahedron Lett. 1989, 30, 5133. b) Deprez, P.; Royer, J.; Husson, H.-P. Synthesis 1991, 759.
- 3. Deprez, P.; Rouden, J.; Chiaroni, A.; Riche, C.; Royer, J.; Husson, H.-P. Tetrahedron Lett. submitted.
- 4. The relative stereochemistry of each cycloadduct was determined by ¹H NMR at 250 MHz on the separated derivatized alcohols obtained after desilylation (citric acid in MeOH).
- 5. The diastereomeric ratio was determined by ¹H and ¹³C NMR on the crude reaction mixture.
- 6. a) Corey, E. J.; Ensley, H. E. J. Am. Chem. Soc. 1975, 97, 6908. b) Oppolzer, W.; Kurth, M.; Reichlin, D.; Chapuis, C.; Mohnhaupt, M.; Moffat, F. Helv. Chim. Acta, 1981, 64, 2802.
- The crude reaction mixture was examined by ¹H NMR (250 MHz) and ¹³C NMR in which the presence of only one product was apparent. HPLC analysis showed small impurities which exhibited UV spectra different of those of cycloadduct.
- 8. **3e** : mp : 134°C ; $[\alpha]_D^{20} = -106$ (c 0.6, CHCl₃); RMN ¹H (CDCl₃, 250 MHz) : 0.1 (s, 9H, TMS), 0.85 (d, J=7Hz, 3H, CH₃), 1.1 (m, 2H, H ment), 1.25 (s, 3H, CH₃), 1.3 (s, 3H, CH₃), 1.5 (m, 4H, H ment), 2.1 (m, 1H, H ment), 2.35 (m, 1H, H ment), 2.65 (t, J=9Hz, 1H, H-5), 2.9 (d, J=7Hz, 1H, H-2), 3.1 (m, 2H, H-3, H-4), 3.55 (d, J=9Hz, 1H, H-5), 3.95 (d, J=6.5Hz, 2H, H-7), 4.35 (t, J=6.5Hz, 1H, H-6), 4.8 (td, J=4.5Hz, 10.5Hz, 1H, H ment), 7.35 (m, 15H, H arom); RMN ¹³C (CDCl₃) : -0.6 (TMS), 21.8 26.1 27.1 31.4 34.7 39.8 40.6 (C ment), 42.6 46.5 50.3 (C ment, C-3, C-4), 50.3 (C-5), 63.2 64.7 (C-2, C-6), 64.6 (C-7), 77.1 (C ment), 124.8 125.4 126.6 127.8 127.9 128.2 128.5 129.0 129.4 132.0 136.1 151.8 (C arom), 168.3 174.8 177.5 (C=O); MS (CI) m/z : 667 (MH⁺, 100), 475 (10), 473 (10); IR (cm⁻¹) : 1715, 1735; Anal. Calcd. for C₄₀H₃₀N₂O₅Si : C 72.03, H 7.35, N 4.20, Found C 71.81, H 7.43, N 4.21.
- 9. **1e**: $[\alpha]_D^{20} = -59$ (c 3.7, CHCl₃); RMN ¹H (CDCl₃, 250 MHz) : 0.90 (d, J=7Hz, 3H, CH₃), 1.25 (s, 3H, CH₃), 1.3 (s, 3H, CH₃), 1.5 (m, 3H, H ment), 1.9 (m, 4H, H ment), 2.1 (m, 1H, H ment), 2.65 (d, J=17Hz, 1H, H-6), 3.0 (d, J=17Hz, 1H, H-6), 3.8 (t, J=7Hz, 1H, H-5), 4.0 (t, J=7Hz, 1H, H-4), 4.3 (t, J=7Hz, 1H, H-5), 4.4 (d, J=3Hz, 1H, H-2), 4.9 (d, J=3Hz, 1H, H-2), 5.0 (td, J=4.5Hz, 10.5Hz, 1H, H ment), 7.35 (m, 10H, H arom) ; RMN ¹³C (CDCl₃) : 21.8 24.4 26.5 28.7 31.3 34.6 39.6 41.8 50.3 (C ment), 52.5 (C-6), 66.4 (C-4), 73.2 (C-5), 74.6 (C ment), 86.6 (C-2), 125.1 125.4 127.6 127.8 128.0 128.7 139.2 151.7 (C arom), 169.8 (C=O) ; MS (CI) m/z : 422 (MH⁺, 100), 302 (5),215 (10) ; IR (cm⁻¹) : 1735 ; Anal. Calcd. for $C_{27}H_{31}NO_3 : C 76.92$, H 8.36, N 3.32, Found : C 76.58, H 8.38, N 3.12.